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By Udo Schwertmann

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Iron oxide - Wikipedia, the free encyclopedia -

Iron oxide pigment. Iron oxides are chemical compounds composed of iron and oxygen. All together, there are sixteen known iron oxides and oxyhydroxides. [1]

https://en.m.wikipedia.org/wiki/Iron_oxide

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http://www.ibrarian.net/navon/paper/Magnetic_iron_oxide_nanoparticles_Synthesis_sta.pdf?paperid=17871683

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<http://nla.gov.au/anbd.bib-an21728034>

Iron Oxide: Lab Chemicals | eBay -

Iron oxide has many uses including pigments, ceramics, welding (thermite), along with a wide range of other uses. As with all chemicals, iron oxide should be handled

<http://www.ebay.com/bhp/iron-oxide>

CiteSeerX Tsourkas A. Superparamagnetic iron -

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<http://academic.research.microsoft.com/Publication/43484331/iron-oxides-in-laboratory>

Iron Oxides | Cosmetics Info -

Iron Oxides used in cosmetic products is an inorganic FDA maintains a laboratory especially for this purpose including thermal decomposition of iron

<http://www.cosmeticsinfo.org/ingredient/iron-oxides>

Iron Oxides - SkinStore.com -

The rust you see on metal is an example of natural iron oxide formation. Rust forms when iron is exposed Iron oxides are made in a lab for safety reasons since

<http://www.skinstore.com/iron-oxides.aspx>

Effects of iron and aluminum oxides and clay -

Eurasian J. Soil Sci. Iron Oxides in the Laboratory: Preparation and characterization, VCH. Schwertmann, U., R.M. Taylor. 1977.

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which may provide insight into ways to produce that form in the lab. Hydrated iron(III) oxides even when they do not contain iron oxide.

[http://en.wikipedia.org/wiki/Iron\(III\)_oxide](http://en.wikipedia.org/wiki/Iron(III)_oxide)